# Emergency Response Guidebook 2012(抜粋)

(U.S. Department of Transportation)

# INTRODUCTION TO GREEN TABLES - INITIAL ISOLATION AND PROTECTIVE ACTION DISTANCES

 Table 1 - Initial Isolation and Protective Action Distances suggests distances useful to protect people from vapors resulting from spills involving dangerous goods that are considered toxic by inhalation (TIH). This list includes certain chemical warfare agents and materials that produce toxic gases upon contact with water. Table 1 provides first responders with initial guidance until technically qualified emergency response personnel are available.

The **Initial Isolation Zone** defines an area SURROUNDING the incident in which persons may be exposed to dangerous (upwind) and life threatening (downwind) concentrations of material. The **Protective Action Zone** defines an area DOWNWIND from the incident in which persons may become incapacitated and unable to take protective action and/or incur serious or irreversible health effects. Table 1 provides specific guidance for small and large spills occurring day or night.

Adjusting distances for a specific incident involves many interdependent variables and should be made only by personnel technically qualified to make such adjustments. For this reason, no precise guidance can be provided in this document to aid in adjusting the table distances; however, general guidance follows.

#### Factors That May Change the Protective Action Distances

**The orange-bordered guide for a material** clearly indicates under the section EVACUATION – Fire, the evacuation distance required to protect against fragmentation hazard of a large container. If the material becomes involved in a **FIRE**, the toxic hazard may be less than the fire or explosion hazard. In these cases, the **Fire** hazard distance should be used.

Initial isolation and protective action distances in this guidebook are derived from historical data on transportation incidents and the use of statistical models. For worst-case scenarios involving the instantaneous release of the entire contents of a package (e.g., as a result of terrorism, sabotage or catastrophic accident) the distances may increase substantially. For such events, doubling of the initial isolation and protective action distances is appropriate in absence of other information.

If more than one tank car containing TIH materials involved in the incident is leaking, LARGE SPILL distances may need to be increased.

For a material with a protective action distance of 11.0+ km (7.0+ miles), the actual distance can be larger in certain atmospheric conditions. If the dangerous goods vapor plume is channeled in a valley or between many tall buildings, distances may be larger than shown in Table 1 due to less mixing of the plume with the atmosphere. Daytime spills in regions with known strong inversions or snow cover, or occurring near sunset, may require an increase of the protective action distance because airborne contaminants mix and disperse more slowly and may travel much farther downwind. In such cases, the nighttime protective action distance may be more appropriate. In addition, protective action distances may be larger for liquid spills when either the material or outdoor temperature exceeds 30°C (86°F).

Materials which react with water to produce large amounts of toxic gases are included in Table 1 - Initial Isolation and Protective Action Distances. Note that some water-reactive materials (WRM) which are also TIH (e.g., Bromine trifluoride (1746), Thionyl chloride (1836), etc.) produce additional TIH materials when spilled in water. For these materials, two entries are provided in Table 1 - Initial Isolation and Protective Action Distances (i.e., for spills on land and for spills in water). If it is not clear whether the spill is on land or in water, or in cases where the spill occurs both on land and in water, choose the larger Protective Action Distance.

Following Table 1, **Table 2** – Water-Reactive Materials Which Produce Toxic Gases lists materials that produce large amounts of Toxic Inhalation Hazard gases (TIH) when spilled in water as well as the toxic gases that are produced when spilled in water.

When a water-reactive TIH producing material is spilled into a river or stream, the source of the toxic gas may move with the current and stretch from the spill point downstream for a substantial distance.

Finally, **Table 3** lists Initial Isolation and Protective Action Distances for Toxic Inhalation Hazard materials that may be more commonly encountered.

The selected materials are:

- Ammonia, anhydrous (UN1005)
- Chlorine (UN1017)
- Ethylene oxide (UN1040)
- Hydrogen chloride (UN1050) and Hydrogen chloride, refrigerated liquid (UN2186)
- Hydrogen fluoride (UN1052)
- Sulfur dioxide/Sulphur dioxide (UN1079)

The materials are presented in alphabetical order and provide Initial Isolation and Protective Action Distances for large spills (more than 208 liters or 55 US gallons) involving different container types (therefore different volume capacities) for day time and night time situations and for different wind speeds.

# PROTECTIVE ACTION DECISION FACTORS TO CONSIDER

The choice of protective actions for a given situation depends on a number of factors. For some cases, evacuation may be the best option; in others, sheltering in-place may be the best course. Sometimes, these two actions may be used in combination. In any emergency, officials need to quickly give the public instructions. The public will need continuing information and instructions while being evacuated or sheltered in-place.

Proper evaluation of the factors listed below will determine the effectiveness of evacuation or in-place protection (shelter in-place). The importance of these factors can vary with emergency conditions. In specific emergencies, other factors may need to be identified and considered as well. This list indicates what kind of information may be needed to make the initial decision.

#### The Dangerous Goods

- Degree of health hazard
- Chemical and physical properties
- Amount involved
- Containment/control of release
- Rate of vapor movement

#### The Population Threatened

- Location
- Number of people
- Time available to evacuate or shelter in-place
- Ability to control evacuation or shelter in-place
- Building types and availability
- Special institutions or populations, e.g., nursing homes, hospitals, prisons

#### Weather Conditions

- Effect on vapor and cloud movement
- Potential for change
- Effect on evacuation or shelter in-place

# PROTECTIVE ACTIONS

**Protective Actions** are those steps taken to preserve the health and safety of emergency responders and the public during an incident involving releases of dangerous goods. Table 1 - Initial Isolation and Protective Action Distances (green-bordered pages) predicts the size of downwind areas which could be affected by a cloud of toxic gas. People in this area should be evacuated and/or sheltered in-place inside buildings.

**Isolate Hazard Area and Deny Entry** means to keep everybody away from the area if they are not directly involved in emergency response operations. Unprotected emergency responders should not be allowed to enter the isolation zone. This "isolation" task is done first to establish control over the area of operations. This is the first step for any protective actions that may follow. See Table 1 - Initial Isolation and Protective Action Distances (green-bordered pages) for more detailed information on specific materials.

**Evacuate** means to move all people from a threatened area to a safer place. To perform an evacuation, there must be enough time for people to be warned, to get ready, and to leave an area. If there is enough time, evacuation is the best protective action. Begin evacuating people nearby and those outdoors in direct view of the scene. When additional help arrives, expand the area to be evacuated downwind and crosswind to at least the extent recommended in this guidebook. Even after people move to the distances recommended, they may not be completely safe from harm. They should not be permitted to congregate at such distances. Send evacuees to a definite place, by a specific route, far enough away so they will not have to be moved again if the wind shifts.

Shelter In-Place means people should seek shelter inside a building and remain inside until the danger passes. Sheltering in-place is used when evacuating the public would cause greater risk than staying where they are, or when an evacuation cannot be performed. Direct the people inside to close all doors and windows and to shut off all ventilating, heating and cooling systems. In-place protection (shelter in-place) may not be the best option if (a) the vapors are flammable; (b) if it will take a long time for the gas to clear the area; or (c) if buildings cannot be closed tightly. Vehicles can offer some protection for a short period if the windows are closed and the ventilating systems are shut off. Vehicles are not as effective as buildings for in-place protection.

It is vital to maintain communications with competent persons inside the building so that they are advised about changing conditions. Persons protected-in-place should be warned to stay far from windows because of the danger from glass and projected metal fragments in a fire and/or explosion.

Every dangerous goods incident is different. Each will have special problems and concerns. Action to protect the public must be selected carefully. These pages can help with **initial** decisions on how to protect the public. Officials must continue to gather information and monitor the situation until the threat is removed.

#### BACKGROUND ON TABLE 1 - INITIAL ISOLATION AND PROTECTIVE ACTION DISTANCES

Initial Isolation and Protective Action Distances in this guidebook were determined for small and large spills occurring during day or night. The overall analysis was statistical in nature and utilized state-of-the-art emission rate and dispersion models; statistical release data from the U.S. DOT HMIS (Hazardous Materials Information System) database; meteorological observations from over 120 locations in United States, Canada and Mexico; and the most current toxicological exposure guidelines.

For each chemical, thousands of hypothetical releases were modeled to account for the statistical variation in both release amount and atmospheric conditions. Based on this statistical sample, the 90<sup>th</sup> percentile Protective Action Distance for each chemical and category was selected to appear in the Table. A brief description of the analysis is provided below. A detailed report outlining the methodology and data used in the generation of the Initial Isolation and Protective Action Distances may be obtained from the U.S. Department of Transportation, Pipeline and Hazardous Materials Safety Administration.

**Release amounts and emission rates** into the atmosphere were statistically modeled based on (1) data from the U.S. DOT HMIS database; (2) container types and sizes authorized for transport as specified in 49 CFR §172.101 and Part 173; (3) physical properties of the individual materials, and (4) atmospheric data from a historical database. The emission model calculated the release of vapor due to evaporation of pools on the ground, direct release of vapors from the container, or a combination of both, as would occur for liquefied gases which can flash to form both a vapor/aerosol mixture and an evaporating pool. In addition, the emission model also calculated the emission of toxic vapor by-products generated from spilling water-reactive materials in water. Spills that involve releases of approximately 208 liters for liquids (55 US gallons) and 300 kg for solids (660 pounds) or less are considered Small Spills, while spills that involve greater quantities are considered Large Spills. An exception to this is certain chemical warfare agents where Small Spills include releases up to 2 kg (4.4 lbs), and Large Spills include releases up to 25 kg (55 lbs). These agents are BZ, CX, GA, GB, GD, GF, HD, HL, HN1, HN2, HN3, L and VX.

**Downwind dispersion** of the vapor was estimated for each case modeled. Atmospheric parameters affecting the dispersion, and the emission rate, were selected in a statistical fashion from a database containing hourly meteorological data from 120 cities in the United States, Canada and Mexico. The dispersion calculation accounted for the time dependent emission rate from the source as well as the density of the vapor plume (i.e., heavy gas effects). Since atmospheric mixing is less effective at dispersing vapor plumes during nighttime, day and night were separated in the analysis. In Table 1, "Day" refers to time periods after sunrise and before sunset, while "Night" includes all hours between sunset and sunrise.

**Toxicological short-term exposure guidelines** for the materials were applied to determine the downwind distance to which persons may become incapacitated and unable to take protective action or may incur serious health effects after a once-in-a-lifetime, or rare, exposure. When available, toxicological exposure guidelines were chosen from AEGL-2 or ERPG-2 emergency response guidelines, with AEGL-2 values being the first choice. For materials that do not have AEGL-2 or ERPG-2 values, emergency response guidelines estimated from lethal concentration limits derived from animal studies were used, as recommended by an independent panel of toxicological experts from industry and academia.

# HOW TO USE TABLE 1 - INITIAL ISOLATION AND PROTECTIVE ACTION DISTANCES

- (1) The responder should already have:
  - Identified the material by its ID Number and Name; (if an ID Number cannot be found, use the Name of Material index in the blue-bordered pages to locate that number.)
  - Found the three-digit guide for that material in order to consult the emergency actions recommended jointly with this table;
  - Noted the wind direction.
- (2) Look in Table 1 (the green-bordered pages) for the ID Number and Name of the Material involved in the incident. Some ID Numbers have more than one shipping name listed look for the specific name of the material. (If the shipping name is not known and Table 1 lists more than one name for the same ID Number, use the entry with the largest protective action distances.)
- (3) Determine if the incident involves a SMALL or LARGE spill and if DAY or NIGHT. Generally, a SMALL SPILL is one which involves a single, small package (e.g., a drum containing up to approximately 208 liters (55 US gallons)), a small cylinder, or a small leak from a large package. A LARGE SPILL is one which involves a spill from a large package, or multiple spills from many small packages. DAY is any time after sunrise and before sunset. NIGHT is any time between sunset and sunrise.
- (4) Look up the INITIAL ISOLATION DISTANCE. Direct all persons to move, in a crosswind direction, away from the spill to the distance specified—in meters and feet.



(5) Look up the initial PROTECTIVE ACTION DISTANCE shown in Table 1. For a given material, spill size, and whether day or night, Table 1 gives the downwind distance—in kilometers and miles— for which protective actions should be considered. For practical purposes, the Protective Action Zone (i.e., the area in which people are at risk of harmful exposure) is a square, whose length and width are the same as the downwind distance shown in Table 1.

(6) Initiate Protective Actions to the extent possible, beginning with those closest to the spill site and working away from the site in the downwind direction. When a water-reactive TIH producing material is spilled into a river or stream, the source of the toxic gas may move with the current or stretch from the spill point downstream for a substantial distance.

The shape of the area in which protective actions should be taken (the Protective Action Zone) is shown in this figure. The spill is located at the center of the small circle. The larger circle represents the INITIAL ISOLATION zone around the spill.



NOTE 1: See "Introduction To Green Tables – Initial Isolation And Protective Action Distances" under "Factors That May Change the Protective Action Distances" (page 285)

#### NOTE 2: See Table 2 – Water-Reactive Materials which Produce Toxic Gases for the list of gases produced when these materials are spilled in water.

Call the emergency response telephone number listed on the shipping paper or the appropriate response agency as soon as possible for additional information on the material, safety precautions and mitigation procedures.

TABLE 1 - INITIAL ISOLATION AND PROTECTIVE ACTION DISTANCES

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ġ	Guide	NAME OF MAIERIAL	Meters	(Feet)	Kilometer	's (Miles)	Kilomete	ers (Miles)	Meters	t (Feet)	Kilomete	ers (Miles)	Kilomete	rs (Miles)
1005 * 1005 *	125 125	Ammonia, anhydrous Anhydrous ammonia	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	150 m	(500 ft)	0.8 km	(0.5 mi)	2.0 km	(1.3 mi)
1008 1008	125 125	Boron trifluoride Boron trifluoride, compressed	30 m	(100 ft)	0.1 km	(0.1 mi)	0.5 km	(0.4 mi)	300 m	(1000 ft)	1.7 km	(1.1 mi)	4.8 km	(3.0 mi)
1016 1016	119 119	Carbon monoxide Carbon monoxide, compressed	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	200 m	(600 ft)	1.2 km	(0.8 mi)	4.8 km	(3.0 mi)
1017 *	124	Chlorine	60 m	(200 ft)	0.4 km	(0.2 mi)	1.5 km	(1.0 mi)	500 m	(1500 ft)	3.0 km	(1.9 mi)	7.9 km	(4.9 mi)
1023 1023	119 119	Coal gas Coal gas, compressed	60 m	(200 ft)	0.2 km	(0.1 mi)	0.2 km	(0.1 mi)	100 m	(300 ft)	0.4 km	(0.2 mi)	0.5 km	(0.3 mi)
1026 1026	119 119	Cyanogen Cyanogen gas	30 m	(100 ft)	0.1 km	(0.1 mi)	0.5 km	(0.3 mi)	60 m	(200 ft)	0.4 km	(0.2 mi)	1.7 km	(1.0 mi)
1040 * 1040 *	119P 119P	Ethylene oxide Ethylene oxide with Nitrogen	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	150 m	(500 ft)	0.9 km	(0.5 mi)	2.0 km	(1.3 mi)
1045 1045	124 124	Fluorine Fluorine, compressed	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	100 m	(300 ft)	0.5 km	(in C.3 mi)	2.3 km	(1.4 mi)
1048	125	Hydrogen bromide, anhydrous	30 m	(100 ft)	0.1 km	(0.1 mi)	0.3 km	(0.2 mi)	200 m	(600 ft)	1.2 km	(0.8 mi)	3.9 km	(2.4 mi)
1050 *	125	Hydrogen chloride, anhydrous	30 m	(100 ft)	0.1 km	(0.1 mi)	0.3 km	(0.2 mi)	60 m	(200 ft)	0.3 km	(0.2 mi)	1.3 km	(0.8 mi)
1051	117	AC (when used as a weapon)	E 09	(200 ft)	0.3 km	(0.2 mi)	1.0 km	(0.6 mi)	E 000	(3000 ft)	3.7 km	(2.3 mi)	8.4 km	(5.3 mi)

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(2.4 mi)		(2.0 ml) (3.5 ml)	(1.2 mi)	(2.0 mi)	(1.7 mi)	(5.9 mi)	(0.3 mi)	(7.0+ mi)	0.3 mi)	(1.5 mi)	(6.7 mi)	(7.0+ mi)	(0.6 mi)	(7.0+ mi)	(0.3 mi)	RIAI
3.8 km		5.6 km	1.9 km	3.2 km	2.7 km	9.5 km	0.5 km	11.0+ km	0.5 km	2.4 km	10.8 km	11.0+ km	0.9 km	11.0+ km	0.5 km	IS MATE
(0.9 mi)	:	(1.0 mi)	(0.4 mi)	(0.7 mi)	(0.7 mi)	(2.3 mi)	(0.2 mi)	(4.7 mi)	(0.2 mi)	(0.7 mi)	(1.9 mi)	(3.5 mi)	(0.3 mi)	(5.8 mi)	(0.2 mi)	FOR TH
1.4 km	-	1.7 km	0.6 km	1.0 km	1.1 km	3.6 km	0.4 km	7.5 km	0.3 km	1.0 km	3.1 km	5.6 km	0.4 km	9.3 km	0.3 km	LABLE 3
(1250 ft)	5	(1000 ft) (1000 ft)	(300 ft)	(500 ft)	(1000 ft)	(2000 ft)	(300 ft)	(3000 ft)	(100 ft)	(600 ft)	(1500 ft)	(3000 ft)	(200 ft)	(2500 ft)	(200 ft)	- T IIISN
400 m	000	300 m	100 m	150 m	300 m	600 m	100 m	1000 m	30 m	200 m	500 m	1000 m	ш 09	m 008	60 m	SO CO
(0.4 mi)	:	(0.3 ml) (0.3 ml)	(0.2 mi)	(0.2 mi)	(0.2 mi)	(0.7 mi)	(0.1 mi)	(2.0 mi)	(0.1 mi)	(0.4 mi)	(1.7 mi)	(1.7 mi)	(0.1 mi)	(2.5 mi)	(0.1 mi)	ASF A
0.6 km	-	0.5 km 0.4 km	0.2 km	0.3 km	0.4 km	1.1 km	0.2 km	3.2 km	0.2 km	0.7 km	2.7 km	2.8 km	0.2 km	4.0 km	0.1 km	+ D
(0.1 mi)		(0.1 ml) (0.1 ml)	(0.1 mi)	(0.1 mi)	(0.1 mi)	(0.2 mi)	(0.1 mi)	(0.5 mi)	(0.1 mi)	(0.1 mi)	(0.4 mi)	(0.4 mi)	(0.1 mi)	(im 6.0)	(0.1 mi)	tions
0.2 km	-	0.1 km	0.1 km	0.1 km	0.1 km	0.2 km	0.2 km	0.8 km	0.2 km	0.2 km	0.6 km	0.7 km	0.1 km	1.4 km	0.1 km	ic condi
(200 ft)	13 0015	(100 ft) (100 ft)	(100 ft)	(100 ft)	(100 ft)	(100 ft)	(200 ft)	(500 ft)	(100 ft)	(100 ft)	(300 ft)	(300 ft)	(100 ft)	(500 ft)	(100 ft)	mospher
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Hydrocyanic acid, aqueous solutions, with more than 20% Hydrogen cyanide Hydrogen cyanide, anhydrous, stabilized	Hydrogen cyanide, stabilized	Hydrogen nuoriae, annyarous Hydrogen sulfide Hydrogen sulphide	Methyl bromide	Methyl mercaptan	Dinitrogen tetroxide Nitrogen dioxide	Nitrosyl chloride	Oil gas Oil gas, compressed	CG (when used as a weapon)	Diphosgene	DP (when used as a weapon)	Phosgene	Sulfur dioxide Sulphur dioxide	Trifluorochloroethylene, stabilized	Acrolein, stabilized	Allyl alcohol	distance can be larger in ce
117	117	117	123	117	124 124	125	119 119	125	125	125	125	125 125	119P	131P	131	means
1051 1051	1051	1053 1053	1062	1064	1067 1067	1069	1071 1071	1076	1076	1076	1076	1079 * 1079 *	1082	1092	1098	4

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₽Ŝ	Gu	uide	NAME OF MATERIAL	Meters	(Feet)	Kilometer	<b>۲</b> s (Miles)	<b>NIC</b> Kilomete	SHT rs (Miles)	Meters	; (Feet)	<b>D</b> . Kilomete	AY rs (Miles)	Kilomete.	<b>àHT</b> rs (Miles)
113	35 13	31	Ethylene chlorohydrin	30 m	(100 ft)	0.1 km	(0.1 mi)	0.1 km	(0.1 mi)	60 m	(200 ft)	0.3 km	(0.2 mi)	0.4 km	(0.3 mi)
114	t3 13 13 13	31P 31P	Crotonaldehyde Crotonaldehyde, stabilized	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	60 m	(200 ft)	0.5 km	(0.3 mi)	1.0 km	(0.6 mi)
116	32 <b>15</b>	55	Dimethyldichlorosilane (when spilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.2 mi)	60 m	(200 ft)	0.6 km	(0.4 mi)	1.9 km	(1.2 mi)
116	33 <b>13</b> 33 <b>13</b>	31	1,1-Dimethylhydrazine Dimethylhydrazine, unsymmetrical	30 m	(100 ft)	0.2 km	(0.1 mi)	0.5 km	(0.4 mi)	100 m	(300 ft)	1.1 km	(0.7 mi)	2.2 km	(1.4 mi)
116	32 15	55	Ethyl chloroformate	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	60 m	(200 ft)	0.4 km	(0.2 mi)	0.6 km	(0.4 mi)
116	33 <b>13</b>	39	Ethyldichlorosilane (when spilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.3 km	(0.2 mi)	60 m	(200 ft)	0.7 km	(0.5 mi)	2.2 km	(1.4 mi)
115	35 13	<b>31P</b>	Ethyleneimine, stabilized	30 m	(100 ft)	0.2 km	(0.1 mi)	0.5 km	(0.3 mi)	100 m	(300 ft)	1.0 km	(0.6 mi)	2.0 km	(1.3 mi)
116	96 15	155	Ethyltrichlorosilane (when spilled in water)	30 m	(100 ft)	0.2 km	(0.1 mi)	0.7 km	(0.5 mi)	200 m	(600 ft)	2.1 km	(1.3 mi)	6.3 km	(3.9 mi)
123	38 15	55	Methyl chloroformate	30 m	(100 ft)	0.2 km	(0.2 mi)	0.6 km	(0.4 mi)	150 m	(500 ft)	1.1 km	(0.7 mi)	2.3 km	(1.4 mi)
123	39 13	31	Methyl chloromethyl ether	30 m	(100 ft)	0.3 km	(0.2 mi)	1.1 km	(0.7 mi)	200 m	(600 ft)	2.2 km	(1.4 mi)	4.6 km	(2.9 mi)
124	42 <b>13</b>	39	Methyldichlorosilane (when spilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.3 km	(0.2 mi)	60 m	(200 ft)	0.8 km	(0.5 mi)	2.5 km	(1.6 mi)
124	44 <b>13</b>	31	Methylhydrazine	30 m	(100 ft)	0.3 km	(0.2 mi)	0.6 km	(0.4 mi)	100 m	(300 ft)	1.4 km	(im 6.0)	2.3 km	(1.4 mi)
12(	50 15	155	Methyltrichlorosilane (when spilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.3 km	(0.2 mi)	100 m	(300 ft)	0.9 km	(0.6 mi)	2.6 km	(1.7 mi)

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1251	131P	Methyl vinyl ketone, stabilized	100 m	(300 ft)	0.3 km	(0.2 mi)	0.8 km	(0.5 mi)	800 m	(2500 ft)	1.5 km	(1.0 mi)	3.0 km	(1.9 mi)
1259	131	Nickel carbonyl	100 m	(300 ft)	1.4 km	(im 6.0)	5.4 km	(3.4 mi)	1000 m	(3000 ft)	11.0+ km	(7.0+ mi)	11.0+ km	(7.0+ mi)
1295	139	Trichlorosilane (when spilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.3 km	(0.2 mi)	ш 09	(200 ft)	0.7 km	(0.4 mi)	2.2 km	(1.4 mi)
1298	155	Trimethylchlorosilane (when spilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	60 m	(200 ft)	0.6 km	(0.4 mi)	1.6 km	(1.0 mi)
1305	155P	Vinyltrichlorosilane (when spilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.2 mi)	60 m	(200 ft)	0.6 km	(0.4 mi)	2.0 km	(1.3 mi)
1305	155P	Vinyltrichlorosilane, stabilized (when spilled in water)												
1340	139	Phosphorus pentasulfide, free from yellow and white Phosphorus (when serilled in water)	30 m	(100 ft)	0.1 km	(0.1 mi)	0.2 km	(0.1 mi)	60 m	(200 ft)	0.4 km	(0.2 mi)	1.4 km	(im (0.0)
1340	139	Photophorus pentasulphide, free from yellow and white Phosphorus (when spilled in water)												
1360	139	Calcium phosphide (when spilled in water)	30 m	(100 ft)	0.2 km	(0.1 mi)	0.7 km	(0.4 mi)	300 m	(1000 ft)	1.1 km	(0.7 mi)	3.8 km	(2.4 mi)
1380	135	Pentaborane	60 m	(200 ft)	0.6 km	(0.4 mi)	2.0 km	(1.2 mi)	200 m	(600 ft)	2.7 km	(1.7 mi)	8.2 km	(5.1 mi)
1384	135	Sodium dithionite	30 m	(100 ft)	0.2 km	(0.1 mi)	0.6 km	(0.4 mi)	60 m	(200 ft)	0.8 km	(0.5 mi)	2.7 km	(1.7 mi)
1384	135	Sodium hydrosulfite (when spilled in water)												
1384	135	Sodium hydrosulphite (when spilled in water)												
1397	139	Aluminum phosphide (when spilled in water)	60 m	(200 ft)	0.2 km	(0.2 mi)	0.9 km	(0.6 mi)	500 m	(1500 ft)	2.1 km	(1.3 mi)	7.5 km	(4.7 mi)
				a successive states of the										

| ≌ Page 295

+" means distance can be larger in certain atmospheric conditions

# HOW TO USE TABLE 2 – WATER-REACTIVE MATERIALS WHICH PRODUCE TOXIC GASES

Table 2 lists materials which produce large amounts of Toxic Inhalation Hazard (TIH) gases when spilled in water and identifies the TIH gases produced.

The materials are listed by ID number order.

These Water Reactive materials are easily identified in Table 1 as their name is immediately followed by (when spilled in water).

Note: Some Water Reactive materials are also TIH materials themselves (e.g., Bromine trifluoride (1746), Thionyl chloride (1836), etc.). In these instances, two entries are provided in **Table 1** for land-based and water-based spills. If the Water Reactive material **is NOT** a TIH and this material **is NOT** spilled in water, **Table 1** and **Table 2** do not apply and safety distances will be found within the appropriate orange guide.

# **TABLE2 - WATER-REACTIVE MATERIALS WHICH PRODUCE TOXIC GASES**

#### Materials Which Produce Large Amounts of Toxic-by-Inhalation (TIH) Gas(es) When Spilled in Water

ID No.	Guide No.	Name of Mater	al			TIH Gas(es) Produced
1162	155	Dimethyldichlorosil	ane			HCI
1183	139	Ethyldichlorosilane				HCI
1196	155	Ethyltrichlorosilane				HCI
1242	139	Methyldichlorosilan	е			HCI
1250	155	Methyltrichlorosilan	е			HCI
1295	139	Trichlorosilane				HCI
1298	155	Trimethylchlorosila	ne			HCI
1305	155P	Vinyltrichlorosilane				HCI
1305	155P	Vinyltrichlorosilane	, stabiliz	ed		HCI
1340	139	Phosphorus pentas	sulfide, f	ree from yellow and white F	hospho	rus H <sub>2</sub> S
1340	139	Phosphorus pentas	ulphide	, free from yellow and white	Phosph	norus H <sub>2</sub> S
1360	139	Calcium phosphide				PH <sub>3</sub>
1384	135	Sodium dithionite				H <sub>s</sub> S SO,
1384	135	Sodium hydrosulfite	Э			H <sub>s</sub> s so,
1384	135	Sodium hydrosulph	ite			H <sub>s</sub> s so,
1397	139	Aluminum phosphic	de			PH3
1419	139	Magnesium alumin	um pho	sphide		PH3
1432	139	Sodium phosphide				PH3
1541	155	Acetone cyanohydr	rin, stab	ilized		HCN
1680	157	Potassium cyanide				HCN
1680	157	Potassium cyanide	, solid			HCN
1689	157	Sodium cyanide				HCN
1689	157	Sodium cyanide, so	olid			HCN
Chemica	al Symb	ols for TIH Gases				
Br <sub>2</sub>	Bromi	ne	HF	Hydrogen fluoride	NO <sub>2</sub>	Nitrogen dioxide
Cl₂ HBr	Chlori Hydro	ne gen bromide	hi H,S	Hydrogen iodide Hydrogen sulfide	PH₃ SO₂	Phosphine Sulfur dioxide
HCI HCN	Hydro Hydro	gen chloride gen cyanide	H₂S NH₃	Hydrogen sulphide Ammonia	SO <sup>2</sup>	Sulphur dioxide

Use this list only when material is spilled in water.

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# **TABLE2 - WATER-REACTIVE MATERIALS WHICH PRODUCE TOXIC GASES**

#### Materials Which Produce Large Amounts of Toxic-by-Inhalation (TIH) Gas(es) When Spilled in Water

ID No.	Guid No.	e Name of Material				TIH ( Proc	Gas(es) luced
1716	156	Acetyl bromide				HBr	
1717	155	Acetyl chloride				HCI	
1724	155	Allyltrichlorosilane, sta	bilized	I		HCI	
1725	137	Aluminum bromide, ar	nhydro	us		HBr	
1726	137	Aluminum chloride, an	hydroi	JS		HCI	
1728	155	Amyltrichlorosilane				HCI	
1732	157	Antimony pentafluorid	е			HF	
1741	125	Boron trichloride				HCI	
1745	144	Bromine pentafluoride				HF	Br <sub>2</sub>
1746	144	Bromine trifluoride				HF	Br <sub>2</sub>
1747	155	Butyltrichlorosilane				HCI	
1752	156	Chloroacetyl chloride				HCI	
1753	156	Chlorophenyltrichloros	silane			HCI	
1754	137	Chlorosulfonic acid				HCI	
1754	137	Chlorosulfonic acid an	d Sulfi	ur trioxide mixture		HCI	
1754	137	Chlorosulphonic acid				HCI	
1754	137	Chlorosulphonic acid a	and Su	Ilphur trioxide mixture		HCI	
1754	137	Sulfur trioxide and Chl	lorosul	fonic acid		HCI	
1754	137	Sulphur trioxide and C	hloros	ulphonic acid		HCI	
1758	137	Chromium oxychloride	)			HCI	
1762	156	Cyclohexenyltrichloros	silane			HCI	
1763	156	Cyclohexyltrichlorosila	ine			HCI	
1765	156	Dichloroacetyl chloride	Э			HCI	
Chemi Br	ical Sy Bro	mbols for TIH Gases:	HF	Hydrogen fluoride	NO	Nitroge	n dioxide
CI <sup>2</sup>	Ch	orine	н	Hydrogen indide		Phoenh	ino

CI,	Chlorine	HI	Hydrogen iodide	PH,	Phosphine
HBr	Hydrogen bromide	H <sub>2</sub> S	Hydrogen sulfide	SO	Sulfur dioxide
HCI	Hydrogen chloride	H,S	Hydrogen sulphide	SO,	Sulphur dioxide
HCN	Hydrogen cyanide	Nĥ₃	Ammonia	2	

Use this list only when material is spilled in water.

# **TABLE2 - WATER-REACTIVE MATERIALS WHICH PRODUCE TOXIC GASES**

#### Materials Which Produce Large Amounts of Toxic-by-Inhalation (TIH) Gas(es) When Spilled in Water

ID No.	Guide No.	e Name of Mate	erial			TIH Gas(es) Produced
1766	156	Dichlorophenyltri	ichlorosila	ne		HCI
1767	155	Diethyldichlorosi	lane			HCI
1769	156	Diphenyldichloro	silane			HCI
1771	156	Dodecyltrichloros	silane			HCI
1777	137	Fluorosulfonic ad	cid			HF
1777	137	Fluorosulphonic	acid			HF
1781	156	Hexadecyltrichlo	rosilane			HCI
1784	156	Hexyltrichlorosila	ane			HCI
1799	156	Nonyltrichlorosila	ane			HCI
1800	156	Octadecyltrichlor	osilane			HCI
1801	156	Octyltrichlorosila	ne			HCI
1804	156	Phenyltrichlorosi	lane			HCI
1806	137	Phosphorus pen	tachloride			HCI
1808	137	Phosphorus tribr	omide			HBr
1809	137	Phosphorus trich	loride			HCI
1810	137	Phosphorus oxy	chloride			HCI
1815	132	Propionyl chlorid	е			HCI
1816	155	Propyltrichlorosil	ane			HCI
1818	157	Silicon tetrachlor	ide			HCI
1828	137	Sulfur chlorides				HCI SO, H,S
1828	137	Sulphur chloride	S			HCI SO, H,S
1834	137	Sulfuryl chloride				HCI
1834	137	Sulphuryl chlorid	е			HCI
Chemica Br <sub>2</sub> Cl <sub>2</sub> HBr HCl	al Symb Brom Chlor Hydro Hydro	ools for TIH Gase ine ogen bromide ogen chloride	es: HF HI H <sub>2</sub> S H <sub>2</sub> S	Hydrogen fluoride Hydrogen iodide Hydrogen sulfide Hydrogen sulphide	NO <sup>2</sup> PH <sup>3</sup> SO <sup>2</sup> <sub>2</sub> SO <sup>2</sup> <sub>2</sub>	Nitrogen dioxide Phosphine Sulfur dioxide Sulphur dioxide

Use this list only when material is spilled in water.

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# HOW TO USE TABLE 3 – INITIAL ISOLATION AND PROTECTIVE ACTION DISTANCES FOR DIFFERENT QUANTITIES OF SIX COMMON TH GASES

Table 3 lists Toxic Inhalation Hazard materials that may be more commonly encountered.

The selected materials are:

- Ammonia (UN1005)
- Chlorine (UN1017)
- Ethylene oxide (UN1040)
- Hydrogen chloride (UN1050) and Hydrogen chloride, refrigerated liquid (UN2186)
- Hydrogen fluoride (UN1052)
- Sulfur dioxide/Sulphur dioxide (UN1079)

The materials are presented in alphabetical order and provide Initial Isolation and Protective Action Distances for large spills (more than 208 liters or 55 US gallons) involving different container types (therefore different volume capacities) for day time and night time situations and different wind speeds.

TABLE	3 - INITI	AL ISOL	ATION	AND P C	ROTEC PF SIX C	TIVE ACTI OMMON 1	on di Tih ga	STANCE	S FOR	DIFFER	ient au	ANTITIES		
	<b>UN</b>	1005 /	\mm(	onia,	anhy	drous: I	_arg	e Spill	S					
	Ϊ	rst ATE				The	en PRO	TECT per:	sons Do	wnwind du	uring			
TRANSPORT CONTAINER		all all				DAY					N	GHT		
			Low (< 6 r ^ 10	wind nph = km/h)	Model (6-12 10 - 2	rate wind 2 mph = 20 km/h)	High (> 12 > 20	n wind mph = km/h)	Low (< 6 - 10	wind mph = km/h)	Modera (6-12 10 - 20	ate wind mph = 0 km/h)	High (> 12 > 20	wind mph = km/h)
	Meters	(Feet)	Кm	(Miles)	К	(Miles)	Кm	(Miles)	К	(Miles)	Km	(Miles)	Кm	(Miles)
Rail tank car	300	(1000)	2.3	(1.4)	1.3	(0.8)	1.0	(9.0)	6.3	(3.9)	2.6	(1.6)	1.3	(0.8)
Highway tank truck or trailer	125	(400)	1.0	(9.0)	0.5	(0.3)	0.3	(0.2)	2.6	(1.6)	0.8	(0.5)	0.5	(0.3)
Agricultural nurse tank	60	(200)	0.6	(0.4)	0.3	(0.2)	0.3	(0.2)	1.5	(6:0)	0.5	(0.3)	0.3	(0.2)
Multiple small cylinders	30	(100)	0.3	(0.2)	0.2	(0.1)	0.2	(0.1)	0.8	(0.5)	0.3	(0.2)	0.2	(0.1)
TRANSPORT CONTAINER	UN	1017 0	Chlor	ine: L	arge	Spills								
Rail tank car	1000	(3000)	11+	(+2)	9.0	(5.6)	5.5	(3.4)	#	(+2)	4	(+2)	7.1	(4.4)
Highway tank truck or trailer	1000	(3000)	10.6	(9.9)	3.5	(2.2)	2.9	(1.8)	ŧ	(+2)	5.5	(3.4)	4.2	(2.6)
Multiple ton cylinders	400	(1250)	4.0	(2.5)	1.5	(6:0)	11	(0.7)	7.9	(4.9)	2.7	(1.7)	1.5	(6.0)
Multiple small cylinders or single ton cylinder	250	(800)	2.6	(1.6)	1.0	(0.6)	0.8	(0.5)	5.6	(3.5)	1.8	(1.1)	0.8	(0.5)

"+" means distance can be larger in certain atmospheric conditions

	Ň	1040 E	Ethyl	ene d	xide:	Large	Spills	<i>(</i> 0							
	Ε	rst ATE				The	n <b>PRO</b>	TECT per	sons Do	wnwind o	during				
TRANSPORT CONTAINER		all all				DAY					Z	IGHT			
			Low (< 6 r < 10	wind nph = km/h)	Moder (6-12 10 - 2	ate wind mph = 0 km/h)	High (> 12 > 20	wind mph = km/h)	Low (< 6 I 10	wind nph = km/h)	Modera (6-12 10 - 20	ate wind mph = 3 km/h)	High v (7) High V 20	n wind mph = km/h)	
	Meters	(Feet)	К	(Miles)	Кm	(Miles)	К	(Miles)	Кm	(Miles)	Кm	(Miles)	Кm	(Miles)	
Rail tank car	200	(009)	1.4	(0.9)	0.8	(0.5)	0.6	(0.4)	4.0	(2.5)	1.4	(0.0)	0.8	(0.5)	
Highway tank truck or trailer	100	(300)	0.8	(0.5)	0.5	(0.3)	0.3	(0.2)	2.1	(1.3)	0.6	(0.4)	0.5	(0.3)	
Multiple small cylinders or single ton cylinder	30	(100)	0.3	(0.2)	0.2	(0.1)	0.2	(0.1)	0.8	(0.5)	0.3	(0.2)	0.2	(0.1)	1
TRANSPORT CONTAINER	N N N N	1050 F 2186 F	Hydro Hydro	ogen ogen	chlori chlori	ide: Laı ide, refi	'ge S 'iger:	pills ated li	iquic	l: Lar	ge Spi	s			
Rail tank car	600	(2000)	6.1	(3.8)	2.3	(1.4)	1.8	(1.1)	ŧ	(+2)	4.0	(2.5)	2.6	(1.6)	
Highway tank truck or trailer	300	(1000)	3.1	(1.9)	÷	(0.7)	0.8	(0.5)	7.4	(4.6)	2.1	(1.3)	1.0	(9.0)	
Multiple ton cylinders	09	(200)	0.6	(0.4)	0.3	(0.2)	0.2	(0.1)	1.8	(1.1)	0.3	(0.2)	0.2	(0.1)	
Multiple small cylinders or single ton cylinder	45	(150)	0.5	(0.3)	0.2	(0.1)	0.2	(0.1)	1.5	(0.9)	0.3	(0.2)	0.2	(0.1)	

	, N	1052 F	lydro	gen	fluori	de: Larç	ge Sp	oills						
	Ε	irst ATE				The	n <b>PROT</b>	ECT per:	sons Downv	vind durinç				
TRANSPORT CONTAINER		all				AY					NIGHT			
		SUOUS	Low (< 6 n < 101	wind nph = km/h)	Moder (6-12 10 - 2	ate wind 2 mph = 10 km/h)	High (> 12 n > 20 k	wind nph = m/h)	Low win (< 6 mph < 10 km/h	0 9 7 W 0	derate win 5-12 mph = 0 - 20 km/h)	н <u>^</u> ^	igh w 12 mp 20 km	ind = hc (h/r
	Meters	(Feet)	Кm	(Miles)	Кm	(Miles)	Km	(Miles)	Km (Mile	s) Kn	n (Miles)	Υ Υ	n (Mi	iles)
Rail tank car	400	(1250)	3.2	(2.0)	1.9	(1.2)	1.6	(1.0)	7.9 (4.9	3.	I (1.9)		9 (1	5
Highway tank truck or trailer	210	(200)	1.9	(1.2)	1.0	(0.6)	0.8	(0.5)	3.9 (2.4	1.0	ŝ (1.0)		0)	(9)
Multiple small cylinders or single ton cylinder	100	(300)	0.8	(0.5)	0.3	(0.2)	0.3	(0.2)	1.6 (1.0	0.1	5 (0.3)	0.	3 (0	.2)
TRANSPORT CONTAINER	NN	1079 S	sulfur	r diox	tide/S	ulphur	dioxi	ide: L	arge S	pills				
Rail tank car	1000	(3000)	11+	(+2)	1+	(+2)	7.6	(4.7)	11+ (7+	11	(+2) +	10	.8 (6	(2)
Highway tank truck or trailer	1000	(3000)	ŧ	(+2)	7.6	(4.7)	5.1	(3.2)	11+ (7+	1	(6.2)	.9	1 (3	(8)
Multiple ton cylinders	600	(2000)	7.1	(4.4)	2.7	(1.7)	1.9	(1.2)	10.5 (6.5	.4.	7 (2.9)		9 (1	.8)
Multiple small cylinders or single ton cylinder	300	(1000)	5.3	(3.3)	1.6	(1.0)	₽	(0.7)	7.9 (4.9	5.	7 (1.7)		5 (0	(6.)

"+" means distance can be larger in certain atmospheric conditions

# ERG2012 USER'S GUIDE

The 2012 Emergency Response Guidebook (ERG2012) was developed jointly by Transport Canada (TC), the U.S. Department of Transportation (DOT), the Secretariat of Transport and Communications of Mexico (SCT) and with the collaboration of CIQUIME (Centro de Información Química para Emergencias) of Argentina, for use by fire fighters, police, and other emergency services personnel who may be the first to arrive at the scene of a transportation incident involving dangerous goods. It is primarily a guide to aid first responders in quickly identifying the specific or generic hazards of the material(s) involved in the incident, and protecting themselves and the general public during the initial response phase of the incident. For the purposes of this guidebook, the "initial response phase" is that period following arrival at the scene of an incident during which the presence and/or identification of dangerous goods is confirmed, protective actions and area securement are initiated, and assistance of qualified personnel is requested. It is not intended to provide information on the physical or chemical properties of dangerous goods.

This guidebook will assist responders in making initial decisions upon arriving at the scene of a dangerous goods incident. It should not be considered as a substitute for emergency response training, knowledge or sound judgment. ERG2012 does not address all possible circumstances that may be associated with a dangerous goods incident. It is primarily designed for use at a dangerous goods incident occurring on a highway or railroad. Be mindful that there may be limited value in its application at fixed facility locations.

ERG2012 incorporates dangerous goods lists from the most recent United Nations Recommendations as well as from other international and national regulations. Explosives are not listed individually by either proper shipping name or ID Number. They do, however, appear under the general heading "Explosives" on the first page of the ID Number index (yellow-bordered pages) and alphabetically in the Name of Material index (blue-bordered pages). Also, the letter (P) following the guide number in the yellow-bordered and blue-bordered pages identifies those materials which present a polymerization hazard under certain conditions, for example: Acrolein, stabilized 131P.

First responders at the scene of a dangerous goods incident should seek additional specific information about any material in question as soon as possible. The information received by contacting the appropriate emergency response agency, by calling the emergency response telephone number on the shipping document, or by consulting the information on or accompanying the shipping document, may be more specific and accurate than this guidebook in providing guidance for the materials involved.

**BEFORE AN EMERGENCY** – **BECOME FAMILIAR WITH THIS GUIDEBOOK!** In the U.S., according to the requirements of the U.S. Department of Labor's Occupational Safety and Health Administration (OSHA, 29 CFR 1910.120), and regulations issued by the U.S. Environmental Protection Agency (EPA, 40 CFR Part 311), first responders must be trained regarding the use of this guidebook.

# **GUIDEBOOK CONTENTS**

**1-Yellow-bordered pages:** Index list of dangerous goods in numerical order of ID number. This section quickly identifies the guide to be consulted from the ID Number of the material involved. This list displays the 4-digit ID number of the material followed by its assigned emergency response guide and the material name.

For example:	ID No.	GUIDE No.	Name of Material
	1090	127	Acetone

**2-Blue-bordered pages:** Index list of dangerous goods in alphabetical order of material name. This section quickly identifies the guide to be consulted from the name of the material involved. This list displays the name of the material followed by its assigned emergency response guide and 4-digit ID number.

For example:	Name of Material	GUIDE No.	ID No.
	Sulfuric acid	137	1830

**3-Orange-bordered pages:** This section is the most important section of the guidebook because it is where all safety recommendations are provided. It comprises a total of 62 individual guides, presented in a two-page format. Each guide provides safety recommendations and emergency response information to protect yourself and the public. The left hand page provides safety related information whereas the right hand page provides emergency response guidance and activities for fire situations, spill or leak incidents and first aid. Each guide is designed to cover a group of materials which possess similar chemical and toxicological characteristics.

The guide title identifies the general hazards of the dangerous goods covered.

For example: GUIDE 124 - Gases-Toxic and/or Corrosive-Oxidizing.

Each guide is divided into three main sections: the first section describes **potential hazards** that the material may display in terms of fire/explosion and health effects upon exposure. The highest potential is listed first. The emergency responder should consult this section first. This allows the responder to make decisions regarding the protection of the emergency response team as well as the surrounding population.

The second section outlines suggested **public safety** measures based on the situation at hand. It provides general information regarding immediate isolation of the incident site, recommended type of protective clothing and respiratory protection. Suggested evacuation distances are listed for small and large spills and for fire situations (fragmentation hazard). It also directs the reader to consult the tables listing Toxic Inhalation Hazard (TIH) materials, chemical warfare agents and water-reactive materials (green-bordered pages) when the material is highlighted in the yellow-bordered and blue-bordered pages.

The third section covers **<u>emergency response</u>** actions, including first aid. It outlines special precautions for incidents which involve fire, spill or chemical exposure. Several recommendations are listed under each part which will further assist in the decision making process. The information on first aid is general guidance prior to seeking medical care.

# 4-Green-bordered pages: This section contains three tables.

Table 1 lists, by ID number order, TIH materials, including certain chemical warfare agents, and water-reactive materials which produce toxic gases upon contact with water. This table provides two different types of recommended safe distances which are "Initial isolation distances" and "Protective action distances". The materials are highlighted in green for easy identification in both numeric (yellow-bordered pages) and alphabetic (blue-bordered pages) lists of the guidebook. This table provides distances for both small (approximately 208 liters (55 US gallons) or less for liquids and 300 kilograms (660 pounds) or less for solids when spilled in water) and large spills (more than 208 liters (55 US gallons) for liquids and more than 300 kilograms (660 pounds) for solids when spilled in water) for all highlighted materials. The list is further subdivided into daytime and nighttime situations. This is necessary due to varying atmospheric conditions which greatly affect the size of the hazardous area. The distances change from daytime to nighttime due to different mixing and dispersion conditions in the air. During the night, the air is generally calmer and this causes the material to disperse less and therefore create a toxic zone which is greater than would usually occur during the day. During the day, a more active atmosphere will cause a greater dispersion of the material resulting in a lower concentration of the material in the surrounding air. The actual area where toxic levels are reached will be smaller (due to increased dispersion). In fact, it is the quantity or concentration of the material vapor that poses problems not its mere presence.

The "Initial Isolation Distance" is a distance within which all persons should be considered for evacuation <u>in all directions</u> from the actual spill/leak source. It is a distance (radius) which defines a circle (Initial Isolation Zone) within which persons may be exposed to dangerous concentrations upwind of the source and may be exposed to life threatening concentrations downwind of the source. For example, in the case of Compressed gas, toxic, n.o.s., ID No. 1955, Inhalation Hazard Zone A, the isolation distance for small spills is 100 meters (300 feet), therefore, representing an evacuation circle of 200 meters (600 feet) in diameter.

For the same material, the "Protective Action Distance" for a small spill is 0.5 kilometers (0.3 mile) for a daytime incident and 2.2 kilometers (1.4 miles) for a nighttime incident, these distances represent a downwind distance from the spill/leak source within which Protective Actions could be implemented. Protective Actions are those steps taken to preserve the health and safety of emergency responders and the public. People in this area could be evacuated and/or sheltered in-place. For more information, consult pages 285 to 291.

<u>What is a TIH?</u> It is a gas or volatile liquid which is known to be so toxic to humans as to pose a hazard to health during transportation, or in the absence of adequate data on human toxicity, is presumed to be toxic to humans because when tested on laboratory animals it has a Lethal Concentration 50 (LC50) value of not more than 5000 ppm.

It is important to note that even though the term zone is used, the hazard zones do not represent any actual area or distance. The assignment of the zones is strictly a function of their Lethal Concentration 50 (LC50); for example, TIH Zone A is more toxic than Zone D. All distances which are listed in the green-bordered pages are calculated by the use of mathematical models for each TIH material. For the assignment of hazard zones refer to the glossary.

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Table 2 lists, by ID number order, materials that produce large amounts of Toxic Inhalation Hazard (TIH) gases when spilled in water and identifies the TIH gases produced. These Water Reactive materials are easily identified in **Table 1** as their name is immediately followed by (**when spilled in water**). Some Water Reactive materials are also TIH materials themselves (e.g., Bromine trifluoride (1746), Thionyl chloride (1836), etc.). In these instances, two entries are provided in **Table 1** for land-based and water-based spills. If the Water Reactive material is NOT a TIH and this material is NOT spilled in water, **Table 1** and **Table 2** do not apply and safety distances will be found within the appropriate orange-bordered guide.

**Table 3** provides, by alphabetical order of material name, initial isolation and protective action distances for six Toxic Inhalation Hazard materials that may be more commonly encountered. The selected materials are:

- Ammonia, anhydrous (UN1005)
- Chlorine (UN1017)
- Ethylene oxide (UN1040)
- Hydrogen chloride (UN1050) and Hydrogen chloride, refrigerated liquid (UN2186)
- Hydrogen fluoride (UN1052)
- Sulfur dioxide/Sulphur dioxide (UN1079)

The table provides Initial Isolation and Protective Action Distances for large spills (more than 208 liters or 55 US gallons) involving different container types (therefore different volume capacities) for day time and night time situations and different wind speeds.

#### **ISOLATION AND EVACUATION DISTANCES**

Isolation or evacuation distances are shown in the guides (orange-bordered pages) and in the Table 1 - Initial Isolation and Protective Action Distances (green-bordered pages). This may confuse users not thoroughly familiar with ERG2012.

It is important to note that some guides refer only to non-TIH materials (36 guides), some refer to both TIH and non-TIH materials (21 guides) and some (5 guides) refer only to TIH or Water-reactive materials (WRM). A guide refers to both TIH and non-TIH materials (for example see GUIDE 131) when the following sentence appears under the title EVACUATION-Spill: "See Table 1 - Initial Isolation and Protective Action Distances for highlighted materials. For non-highlighted materials, increase, in the downwind direction, as necessary, the isolation distance shown under 'PUBLIC SAFETY.'" A guide refers only to TIH or WRM materials (for example see GUIDE 124) when the following sentence appears under the title EVACUATION-Spill: "See Table 1 - Initial Isolation and Protective Action Distances". If the previous sentences do not appear in a guide, then this particular guide refers only to non-TIH materials (for example see GUIDE 128).

In order to identify appropriate isolation and protective action distances, use the following:

If you are dealing with a **TIH/WRM/Chemical warfare** material (highlighted entries in the index lists), the isolation and evacuation distances are found directly in the green-bordered pages. The guides (orange-bordered pages) also remind the user to refer to the green-bordered pages for evacuation specific information involving highlighted materials.

If you are dealing with a **non-TIH material but the guide refers to both TIH and non-TIH materials**, an immediate isolation distance is provided under the heading PUBLIC SAFETY as a precautionary measure to prevent injuries. It applies to the non-TIH materials only. In addition, for evacuation purposes, the guide informs the user under the title EVACUATION-Spill to increase, for non-highlighted materials, in the downwind direction, if necessary, the immediate isolation distance listed under "PUBLIC SAFETY". For example, GUIDE 131 – Flammable Liquids-Toxic, instructs the user to: "As an immediate precautionary measure, isolate spill or leak area for at least 50 meters (150 feet) in all directions." In case of a large spill, the isolation area could be expanded from 50 meters (150 feet) to a distance deemed as safe by the On-scene commander and emergency responders.

If you are dealing with a **non-TIH material and the guide refers only to non-TIH materials**, the immediate isolation and evacuation distances are specified as actual distances in the guide (orange-bordered pages) and are not referenced in the green-bordered pages.

- Note 1: If an entry is highlighted in green in either the yellow-bordered or blue-bordered pages AND THERE IS NO FIRE, go directly to Table 1 Initial Isolation and Protective Action Distances (green-bordered pages) and look up the ID number and name of material to obtain initial isolation and protective action distances. IF THERE IS A FIRE, or IF A FIRE IS INVOLVED, ALSO CONSULT the assigned guide (orange-bordered pages) and apply as appropriate the evacuation information shown under PUBLIC SAFETY.
- Note 2: If the name in Table 1 is shown with "When Spilled In Water", these materials produce large amounts of Toxic Inhalation Hazard (TIH) gases when spilled in water. Some Water Reactive materials are also TIH materials themselves (e.g., Bromine trifluoride (1746), Thionyl chloride (1836), etc.). In these instances, two entries are provided in Table 1 for land-based and water-based spills. If the Water Reactive material is NOT a TIH and this material is NOT spilled in water, Table 1 and Table 2 do not apply and safety distances will be found within the appropriate orange-bordered guide.

# **PROTECTIVE CLOTHING**

**Street Clothing and Work Uniforms.** These garments, such as uniforms worn by police and emergency medical services personnel, provide almost no protection from the harmful effects of dangerous goods.

Structural Fire Fighters' Protective Clothing (SFPC). This category of clothing, often called turnout or bunker gear, means the protective clothing normally worn by fire fighters during structural fire fighting operations. It includes a helmet, coat, pants, boots, gloves and a hood to cover parts of the head not protected by the helmet and facepiece. This clothing must be used with full-facepiece positive pressure self-contained breathing apparatus (SCBA). This protective clothing should, at a minimum, meet the OSHA Fire Brigades Standard (29 CFR 1910.156). Structural fire fighters' protective clothing provides limited protection from heat and cold, but may not provide adequate protection from the harmful vapors or liquids that are encountered during dangerous goods incidents. Each guide includes a statement about the use of SFPC in incidents involving those materials referenced by that guide. Some guides state that SFPC provides limited protection. In those cases, the responder wearing SFPC and SCBA may be able to perform an expedient, that is quick "in-and-out", operation. However, this type of operation can place the responder at risk of exposure, injury or death. The incident commander makes the decision to perform this operation only if an overriding benefit can be gained (i.e., perform an immediate rescue, turn off a valve to control a leak, etc.). The coverall-type protective clothing customarily worn to fight fires in forests or wildlands is not SFPC and is not recommended nor referred to elsewhere in this guidebook.

**Positive Pressure Self-Contained Breathing Apparatus (SCBA)**. This apparatus provides a constant, positive pressure flow of air within the facepiece, even if one inhales deeply while doing heavy work. Use apparatus certified by NIOSH and the Department of Labor/Mine Safety and Health Administration in accordance with 42 CFR Part 84. Use it in accordance with the requirements for respiratory protection specified in OSHA 29 CFR 1910.134 (Respiratory Protection) and/or 29 CFR 1910.156 (f) (Fire Brigades Standard). Chemical-cartridge respirators or other filtering masks are not acceptable substitutes for positive pressure self-contained breathing apparatus. Demand-type SCBA does not meet the OSHA 29 CFR 1910.156 (f)(1)(i) of the Fire Brigades Standard. If it is suspected that a Chemical Warfare Agent (CW) is involved, the use of NIOSH-certified respirators with CBRN protection are highly recommended.

**Chemical Protective Clothing and Equipment**. Safe use of this type of protective clothing and equipment requires specific skills developed through training and experience. It is generally not available to, or used by, first responders. This type of special clothing may protect against one chemical, yet be readily permeated by chemicals for which it was not designed. Therefore, protective clothing should not be used unless it is compatible with the released material. This type of special clothing offers little or no protection against heat and/ or cold. Examples of this type of equipment have been described as (1) Vapor Protective Suits (NFPA 1991), also known as Totally-Encapsulating Chemical Protective (TECP) Suits or Level A\* protection (OSHA 29 CFR 1910.120, Appendix A & B), and (2) Liquid-Splash

Protective Suits (NFPA 1992 & 1993), also known as Level B\* or C\* protection (OSHA 29 CFR 1910.120, Appendix A & B) or suits for chemical/biological terrorism incidents (NFPA 1994), class 1, 2 or 3 Ensembles and Standard CAN/CGSB/CSA-Z1610-11 – Protection of first responders from chemical, biological, radiological, and nuclear (CBRN) events (2011). No single protective clothing material will protect you from all dangerous goods. Do not assume any protective clothing is resistant to cold and/or heat or flame exposure unless it is so certified by the manufacturer (NFPA 1991 5-3 Flammability Resistance Test and 5-6 Cold Temperature Performance Test).

\* Consult glossary for additional protection levels under the heading "Protective Clothing".

# FIRE AND SPILL CONTROL

#### FIRE CONTROL

Water is the most common and generally most available fire extinguishing agent. Exercise caution in selecting a fire extinguishing method since there are many factors to be considered in an incident. Water may be ineffective in fighting fires involving some materials; its effectiveness depends greatly on the method of application.

Fires involving a spill of flammable liquids are generally controlled by applying a fire fighting foam to the surface of the burning material. Fighting flammable liquid fires requires foam concentrate which is chemically compatible with the burning material, correct mixing of the foam concentrate with water and air, and careful application and maintenance of the foam blanket. There are two general types of fire fighting foam: regular and alcohol-resistant. Examples of regular foam are protein-base, fluoroprotein, and aqueous film forming foam (AFFF). Some flammable liquids, including many petroleum products, can be controlled by applying regular foam. Other flammable liquids, including polar solvents (flammable liquids which are water soluble) such as alcohols and ketones, have different chemical properties. A fire involving these materials cannot be easily controlled with regular foam and requires application of alcohol-resistant foam. Polar-solvent fires may be difficult to control and require a higher foam application rate than other flammable liquid fires (see NFPA/ANSI Standards 11 and 11A for further information). Refer to the appropriate guide to determine which type of foam is recommended. Although it is impossible to make specific recommendations for flammable liquids which have subsidiary corrosive or toxic hazards, alcohol-resistant foam may be effective for many of these materials. The emergency response telephone number on the shipping document, or the appropriate emergency response agency, should be contacted as soon as possible for guidance on the proper fire extinguishing agent to use. The final selection of the agent and method depends on many factors such as incident location, exposure hazards, size of the fire, environmental concerns, as well as the availability of extinguishing agents and equipment at the scene.

#### WATER REACTIVE MATERIALS

Water is sometimes used to flush spills and to reduce or direct vapors in spill situations. Some of the materials covered by the guidebook can react violently or even explosively with water. In these cases, consider letting the fire burn or leaving the spill alone (except to prevent its spreading by diking) until additional technical advice can be obtained. The applicable guides clearly warn you of these potentially dangerous reactions. These materials require technical advice since

- (1) water getting inside a ruptured or leaking container may cause an explosion;
- (2) water may be needed to cool adjoining containers to prevent their rupturing (exploding) or further spread of the fires;

- (3) water may be effective in mitigating an incident involving a water-reactive material only if it can be applied at a sufficient flooding rate for an extended period; and
- (4) the products from the reaction with water may be more toxic, corrosive, or otherwise more undesirable than the product of the fire without water applied.

When responding to an incident involving water-reactive materials, take into account the existing conditions such as wind, precipitation, location and accessibility to the incident, as well as the availability of the agents to control the fire or spill. Because there are variables to consider, the decision to use water on fires or spills involving water-reactive materials should be based on information from an authoritative source; for example, a producer of the material, who can be contacted through the emergency response telephone number or the appropriate emergency response agency.

# VAPOR CONTROL

Limiting the amount of vapor released from a pool of flammable or corrosive liquids is an operational concern. It requires the use of proper protective clothing, specialized equipment, appropriate chemical agents, and skilled personnel. Before engaging in vapor control, get advice from an authoritative source as to the proper tactics.

There are several ways to minimize the amount of vapors escaping from pools of spilled liquids, such as special foams, adsorbing agents, absorbing agents, and neutralizing agents. To be effective, these vapor control methods must be selected for the specific material involved and performed in a manner that will mitigate, not worsen, the incident.

Where specific materials are known, such as at manufacturing or storage facilities, it is desirable for the dangerous goods response team to prearrange with the facility operators to select and stockpile these control agents in advance of a spill. In the field, first responders may not have the most effective vapor control agent for the material available. They are likely to have only water and only one type of fire fighting foam on their vehicles. If the available foam is inappropriate for use, they are likely to use water spray. Because the water is being used to form a vapor seal, care must be taken not to churn or further spread the spill during application. Vapors that do not react with water may be directed away from the site using the air currents surrounding the water spray. Before using water spray or other methods to safely control vapor emission or to suppress ignition, obtain technical advice, based on specific chemical name identification.

# BLEVE (Boiling Liquid Expanding Vapor Explosion)

The following section presents, in a two-page format, background information on BLEVEs and includes a chart that provides important safety-related information to consider when confronted with this type of situation involving Liquefied Petroleum Gases (LPG), UN1075. LPGs include the following flammable gases; Butane, UN1011; Butylene, UN1012; Isobutylene, UN1055; Propylene, UN2077; Isobutane, UN1969; and Propane, UN1978.

### What are the main hazards from a BLEVE?

The main hazards from a propane or LPG BLEVE are:

- fire
- thermal radiation from the fire
- blast
- projectiles

The danger from these decreases as you move away from the BLEVE centre. The furthest reaching hazard is projectiles.

This information was prepared for Transport Canada, the Canadian Association of Fire Chiefs and the Propane Gas Association of Canada Inc. by Dr. A. M. Birk, Queen's University, Kingston (Ontario) Canada.

For a free download or to order a DVD of the video **BLEVE Response and Prevention**, please visit <<u>http://www.tc.gc.ca/eng/tdg/bleve-1119.htm</u>> or contact us at 1-888-830-4911, or by Email: MPS@tc.gc.ca.

To download a free copy, first click on the green "View/Download" button and then left-click the video link to view the video or right-click to download a copy by selecting "Save target as" to save to your computer.

# **BLEVE – SAFETY PRECAUTIONS**

**Use with caution**. The following table gives a summary of tank properties, critical times, critical distances and cooling water flow rates for various tank sizes. This table is provided to give responders some guidance but it should be used with caution.

Tank dimensions are approximate and can vary depending on the tank design and application.

**Minimum time to failure** is based on *severe torch fire impingement* on the vapour space of a tank in good condition, and is approximate. Tanks may fail earlier if they are damaged or corroded. Tanks may fail minutes or hours later than these minimum times depending on the conditions. It has been assumed here that the tanks are not equipped with thermal barriers or water spray cooling.

**Minimum time to empty** is based on an engulfing fire with a properly sized pressure relief valve. If the tank is only partially engulfed then time to empty will increase (i.e., if tank is 50% engulfed then the tanks will take twice as long to empty). Once again, it has been assumed that the tank is not equipped with a thermal barrier or water spray.

Tanks equipped with thermal barriers or water spray cooling significantly increase the times to failure and the times to empty. A thermal barrier can reduce the heat input to a tank by a factor of ten or more. This means it could take ten times as long to empty the tank through the Pressure Relief Valve (PRV).

**Fireball radius and emergency response distance** is based on mathematical equations and is approximate. They assume spherical fireballs and this is not always the case.

**Two safety distances for public evacuation**. The minimum distance is based on tanks that are launched with a small elevation angle (i.e., a few degrees above horizontal). This is most common for horizontal cylinders. The preferred evacuation distance has more margin of safety since it assumes the tanks are launched at a 45 degree angle to the horizontal. This might be more appropriate if a vertical cylinder is involved.

It is understood that these distances are very large and may not be practical in a highly populated area. However, it should be understood that the risks increase rapidly the closer you are to a BLEVE. Keep in mind that the furthest reaching projectiles tend to come off in the zones 45 degrees on each side of the tank ends.

Water flow rate is based on  $\sqrt[5]{capacity (USgal)}$  = usgal/min needed to cool tank metal.

**Warning**: the data given are approximate and should only be used with extreme caution. For example, where times are given for tank failure or tank emptying through the pressure relief valve – these times are typical but they can vary from situation to situation. Therefore, never risk life based on these times.

	water ate	Sgal/min	25	20	112	158	224	371	512	716	935
	Cooling v flow ra	tres/min U	94.6	189.3	424	598	848	1404	1938	2710	3539
	ation	(Feet) Li	(1007)	(1601)	(2736)	(3445)	(4341)	(6076)	(7218)	(7218)	(7218)
	Prefel evacuá distal	Meters	307	488	834	1050	1323	1852	2200	2200	2200
	ation	(Feet)	(505)	(801)	(1368)	(1722)	(2169)	(3038)	(0770)	(4708)	(5627)
	Minim evacua dista	Meters	154	244	417 (	525 (	661 (	926	1149 (	1435 (	1715 (
	ency nse nce	(Feet)	(295)	(295)	(364)	(459)	(577)	(810)	(1004)	(1257)	(1499)
	Emerge respo distar	Meters	6	06	Ŧ	140	176	247	306	383 (	457 (
	us us	(Feet)	(33)	(53)	(92)	(115)	(144)	(203)	(253)	(315)	(374)
TION	Firel	Meters	<b>e</b>	16	28	35	44	62	1	96	114
BLEVE WITH CAU	Approximate time to empty for engulfing fire	Minutes	8	12	18	20	22	28	32	40	45
(USE	Minimum time to failure for severe torch	Minutes	4	4	ъ	Ω	G	7	7	80	6
	ane ss ns(Lbs)	ls(Lbs)	(88)	(353)	(1764)	(3527)	(7055)	19400)	37037)	72310)	23457)
	Prop Mas	Kilogran	40	160	800	1600	3200	8800	16800 (	32800 (	56000 (1
	gth	(Feet)	(4.9)	(4.9)	(8.6)	(16.1)	(21.3)	(22)	(38.7)	(45)	(56.4)
_	Len	Meters	1.5	1.5	ę	4.9	6.5	6.7	11.8	13.7	17.2
	eter	(Feet)	(1)	(2)	(3.2)	(3.3)	(4.1)	(6.9)	(6.9)	(6)	(10.8)
	Diam	Meters	0.3	0.61	96.0	-	1.25	2.1	2.1	2.75	3.3
	acity	(Gallons)	(38.6)	(154.4)	(772)	(1544)	(3088)	(8492)	(16212)	(31652)	(54040)
	Capi	Litres (	<del>1</del> 0	400	2000	4000	8000	22000	42000	82000	140000

# CRIMINAL/TERRORIST USE OF CHEMICAL/BIOLOGICAL/RADIOLOGICAL AGENTS

The following is intended to supply information to first responders for use in making a preliminary assessment of a situation that they suspect involves criminal/terrorist use of chemical, biological agents and/or radioactive materials (CBRN). To aid in the assessment, a list of observable indicators of the use and/or presence of a CB agent or radioactive material is provided in the following paragraphs. This section ends with a Safe Standoff Distance Chart for various threats when Improvised Explosive Devices are involved.

# DIFFERENCES BETWEEN A CHEMICAL, BIOLOGICAL AND RADIOLOGICAL AGENT

Chemical and biological agents as well as radioactive materials can be dispersed in the air we breathe, the water we drink, or on surfaces we physically contact. Dispersion methods may be as simple as opening a container, using conventional (garden) spray devices, or as elaborate as detonating an improvised explosive device.

**Chemical Incidents** are characterized by the rapid onset of medical symptoms (minutes to hours) and easily observed signatures (colored residue, dead foliage, pungent odor, dead insects and animals).

**Biological Incidents** are characterized by the onset of symptoms in hours to days. Typically, there will be no characteristic signatures because biological agents are usually odorless and colorless. Because of the delayed onset of symptoms in a biological incident, the area affected may be greater due to the movement of infected individuals.

**Radiological Incidents** are characterized by the onset of symptoms, if any, in days to weeks or longer. Typically, there will be no characteristic signatures because radioactive materials are usually odorless and colorless. Specialized equipment is required to determine the size of the affected area, and whether the level of radioactivity presents an immediate or long-term health hazard. Because radioactivity is not detectable without special equipment, the affected area may be greater due to the migration of contaminated individuals.

At the levels created by most probable sources, not enough radiation would be generated to kill people or cause severe illness. In a radiological incident generated by a "dirty bomb", or Radiological Dispersal Device (RDD), in which a conventional explosive is detonated to spread radioactive contamination, the primary hazard is from the explosion. However, certain radioactive materials dispersed in the air could contaminate up to several city blocks, creating fear and possibly panic, and requiring potentially costly cleanup.

#### INDICATORS OF A POSSIBLE CHEMICAL INCIDENT

Dead animals/birds/fish	Not just an occasional road kill, but numerous animals (wild and domestic, small and large), birds, and fish in the same area.
Lack of insect life	If normal insect activity (ground, air, and/or water) is missing, check the ground/water surface/shore line for dead insects. If near water, check for dead fish/aquatic birds.

#### •

INDICATORS OF A POSSIBLE CF Unexplained odors	Smells may range from fruity to flowery to sharp/pungent to garlic/ horseradish-like to bitter almonds/peach kernels
	to new mown hay. It is important to note that the particular odor is completely out of character with its surroundings.
Unusual numbers of dying or sick people (mass casualties)	Health problems including nausea, disorientation, difficulty in breathing, convulsions, localized sweating, conjunctivitis (reddening of eyes/nerve agent symptoms), erythema (reddening of skin/vesicant symptoms) and death.
Pattern of casualties	Casualties will likely be distributed downwind, or if indoors, by the air ventilation system.
Blisters/rashes	Numerous individuals experiencing unexplained water-like blisters, weals (like bee stings), and/or rashes.
Illness in confined area	Different casualty rates for people working indoors versus outdoors dependent on where the agent was released.
Unusual liquid droplets	Numerous surfaces exhibit oily droplets/film; numerous water surfaces have an oily film. (No recent rain.)
Different looking areas	Not just a patch of dead weeds, but trees, shrubs, bushes, food crops, and/or lawns that are dead, discolored, or withered. (No current drought.)
Low-lying clouds	Low-lying cloud/fog-like condition that is not consistent with its surroundings.
Unusual metal debris	Unexplained bomb/munitions-like material, especially if it contains a liquid.
INDICATORS OF A POSSIBLE BI	OLOGICAL INCIDENT
Unusual numbers of sick or dying people or animals	Any number of symptoms may occur. Casualties may occur hours to days after an incident has occurred. The time required before symptoms are observed is dependent on the agent used.
Unscheduled and unusual spray being disseminated	Especially if outdoors during periods of darkness.
Abandoned spray devices	Devices may not have distinct odors.
INDICATORS OF A POSSIBLE RA	ADIOLOGICAL INCIDENT
Radiation Symbols	Containers may display a "propeller" radiation symbol.

Unusual metal debris

Unexplained bomb/munitions-like material.

# INDICATORS OF A POSSIBLE RADIOLOGICAL INCIDENT (continued)

Heat-emitting material	Material that is hot or seems to emit heat without any sign of an external heat source.
Glowing material	Strongly radioactive material may emit or cause radioluminescence.
Sick people/animals	In very improbable scenarios there may be unusual numbers of sick or dying people or animals. Casualties may occur hours to days or weeks after an incident has occurred. The time required before symptoms are observed is dependent on the radioactive material used, and the dose received. Possible symptoms include skin reddening or vomiting.

# PERSONAL SAFETY CONSIDERATIONS

When approaching a scene that may involve CB agents or radioactive materials, the most critical consideration is the safety of oneself and other responders. Protective clothing and respiratory protection of appropriate level of safety must be used. In incidents where it is suspected that CBRN materials have been used as weapons, NIOSH-certified respirators with CBRN protection are highly recommended. Be aware that the presence and identification of CB agents or radioactive materials may not be verifiable, especially in the case of biological or radiological agents. The following actions/measures to be considered are applicable to either a chemical, biological or radiological incident. The guidance is general in nature, not all encompassing, and its applicability should be evaluated on a case-by-case basis.

**Approach and response strategies.** Protect yourself and use a safe approach (minimize any exposure time, maximize the distance between you and the item that is likely to harm you, use cover as protection and wear appropriate personal protective equipment and respiratory protection). Identify and estimate the hazard by using indicators as provided above. Isolate the area and secure the scene; potentially contaminated people should be isolated and decontaminated as soon as possible. To the extent possible, take measures to limit the spread of contamination. In the event of a chemical incident, the fading of chemical odors is not necessarily an indication of reduced vapor concentrations. Some chemicals deaden the senses giving the false perception that the chemical is no longer present.

If there is any indication that an area may be contaminated with radioactive materials, including the site of any non-accidental explosion, responder personnel should be equipped with radiation detection equipment that would alert them if they are entering a radiologically compromised environment, and should have received adequate training in its use. This equipment should be designed in such a way that it can also alert the responders when an unacceptable ambient dose rate or ambient dose has been reached.

Initial actions to consider in a potential CBRN/Hazmat Terrorism Event:

- Avoid using cell phones, radios, etc. within 100 meters (300 feet) of a suspect device
- NOTIFY your local police by calling 911.
- Set up Incident command upwind and uphill of the area.
- Do NOT touch or move suspicious packages/containers.
- Be cautious regarding potential presence of secondary devices (e.g. Improvised Explosive Devices, IEDs).
- Avoid contamination.
- Limit access to only those responsible for rescue of victims or assessment of unknown materials or devices.
- Evacuate and isolate individuals potentially exposed to dangerous goods/hazardous materials.
- · Isolate contaminated areas and secure the scene for analysis of material.

**Decontamination measures.** Emergency responders should follow standard decontamination procedures (flush-strip-flush). Mass casualty decontamination should begin as soon as possible by stripping (all clothing) and flushing (soap and water). If biological agents are involved or suspected, careful washing and use of a brush are more effective. If chemical agents are suspected, the most important and effective decontamination should be performed using a 0.5% hypochlorite solution (1 part household bleach mixed with 9 parts water). If biological agents are suspected, a contact time of 10 to 15 minutes should be allowed before rinsing. The solution can be used on soft tissue wounds, but must not be used in eyes or open wounds of the abdomen, chest, head, or spine. For further information contact the agencies listed in this guidebook.

For persons contaminated with radioactive material, remove them to a low radiation area if necessary. Remove their clothing and place it in a clearly marked sealed receptacle, such as a plastic bag, for later testing. Use decontamination methods described above, but avoid breaking the skin, e.g., from shaving, or overly vigorous brushing. External radiological contamination on intact skin surface rarely causes a high enough dose to be a hazard to either the contaminated person or the first responders. For this reason, except in very unusual circumstances, an injured person who is also radiologically contaminated should be medically stabilized, taking care to minimize the spread of the contamination to the extent possible, before decontamination measures are initiated.

**Note:** The above information was developed in part by the Department of National Defence (Canada), the U.S. Department of the Army, Aberdeen Proving Ground and the Federal Bureau of Investigation (FBI).

#### Improvised Explosive Device (IED) SAFE STAND OFF DISTANCE

	Threat Description	Explosives Mass (TNT equivalent) <sup>1</sup>		Building Evacuation Distance <sup>2</sup>		Outdoor Evacuation Distance <sup>3</sup>		
	Pipe Bomb	5 lbs	2.3 kg	70 ft	21 m	850 ft	259 m	
int)	Suicide Belt	10 lbs	4.5 kg	90 ft	27 m	1,080 ft	330 m	
ivale	Suicide Vest	20 lbs	9 kg	110 ft	34 m	1,360 ft	415 m	
.Equ	Briefcase/Suitcase Bomb	50 lbs	23 kg	150 ft	46 m	1,850 ft	564 m	
LNT)	Compact Sedan	500 lbs	227 kg	320 ft	98 m	1,500 ft	457 m	
sives	Sedan	1,000 lbs	454 kg	400 ft	122 m	1,750 ft	534 m	
xplo	Passenger/Cargo Van	4,000 lbs	1 814 kg	640 ft	195 m	2,750 ft	838 m	
High E	Small Moving Van/ Delivery Truck	10,000 lbs	4 536 kg	860 ft	263 m	3,750 ft	1 143 m	
	Moving Van/Water Truck	30,000 lbs	13 608 kg	1,240 ft	375 m	6,500 ft	1 982 m	
	Semitrailer	60,000 lbs	27 216 kg	1,570 ft	475 m	7,000 ft	2 134 m	

	Threat Description	LPG N Volu	lass/ me¹	Firel Diame	oall eter⁴	Safe Distance⁵	
Gas ane)	Small LPG Tank	20 lbs/5 gal	9 kg/19 L	40 ft	12 m	160 ft	48 m
eum ( Prop	Large LPG Tank	100 lbs/25 gal	45 kg/95 L	69 ft	21 m	276 ft	84 m
Petrol tane or	Commercial/ Residential LPG Tank	2,000 lbs/500 gal	907 kg/1 893 L	184 ft	56 m	736 ft	224 m
uefied à - But	Small LPG Truck	8,000 lbs/2,000 gal	3 630 kg/7 570 L	292 ft	89 m	1,168 ft	356 m
Liqi (LPG	Semitanker LPG	40,000 lbs/10,000 gal	18 144 kg/37 850 L	499 ft	152 m	1,996 ft	608 m

<sup>1</sup> Based on the maximum amount of material that could reasonably fit into a container or vehicle. Variations possible.

<sup>2</sup> Governed by the ability of an unreinforced building to withstand severe damage or collapse.

<sup>3</sup> Governed by the greater of fragment throw distance or glass breakage/falling glass hazard distance. These distances can be reduced for personnel wearing ballistic protection. Note that the pipe bomb, suicide belt/vest, and briefcase/ suitcase bomb are assumed to have a fragmentation characteristic that requires greater standoff distances than an equal amount of explosives in a vehicle.

<sup>4</sup> Assuming efficient mixing of the flammable gas with ambient air.

<sup>5</sup> Determined by U.S. firefighting practices wherein safe distances are approximately 4 times the flame height. Note that an LPG tank filled with high explosives would require a significantly greater standoff distance than if it were filled with LPG.